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london dispersion force dipole dipole force hydrogen

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26 dec 2019 london dispersion forces are generally weaker than other types of intermolecular forces such as hydrogen bonds or dipole dipole interactions however they play an important role in

toroidal dipole

intermolecularforce londondispersion forces dipole dipoleforces hbonding polarizability polarity

the term van der waals force is sometimes used loosely for all intermolecular forces 3 the term always includes the london dispersion force between instantaneously induced dipoles citation

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